



Temperature and composition dependence of N₂O₅ uptake coefficients

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Nitrogen oxides play a central role in the chemistry of the atmosphere. Heterogeneous removal of N₂O₅ can be a major loss route for NO_x with modelling work by Tie et al (2003) suggesting that, at high latitudes, N₂O₅ hydrolysis can reduce NO_x levels by as much as 90 %.

The majority of laboratory studies of the heterogeneous uptake of N₂O₅ by aerosol particles have been performed at room temperature. At present, the role of temperature is the largest source of uncertainty in global loss rates. We are undertaking a series of laboratory measurements of the rate of removal of N₂O₅ by aerosol particles using a new apparatus capable of reaching temperatures relevant to the polar troposphere.

These complement our earlier measurements on sulfates (Hallquist et al., 2003) and sulfates mixed with high molecular weight organics (Badger et al., 2006).

The instrument has been developed in conjunction with Tapcon, Austria, and comprises a temperature-controlled atmospheric pressure aerosol flow tube and differential mobility analyser. The operating range of the instrument is 240-320 K, with control of relative humidity over the range 0-80%.

We present measurements of the uptake coefficient, γ , for loss of N₂O₅ made using the new apparatus, together with an analysis of the data in terms of particle composition, size and other parameters. The development of a kinetic model for uptake, employing explicit treatment of the coupled processes of reaction and diffusion in the aerosol particle will be described.