



## **CO<sub>2</sub> concentration in soil air affects recrystallisation rate of primary CaCO<sub>3</sub>**

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Pedogenic (secondary) carbonates originate from dissolution and recrystallisation of lithogenic (primary) carbonates with CO<sub>2</sub> from soil air, leading to a complete exchange of the lithogenic with the atmospheric carbon (C) during time. Therefore, isotopic signature of pedogenic carbonates ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$ ) will be used as important information for reconstruction of paleoclimatic conditions. Therefore, the interest in pedogenic carbonates increased over the last years. However, the recrystallisation rate of primary CaCO<sub>3</sub> by pedogenic carbonate formation and the dependence of the rate on environmental factors (i.e. CO<sub>2</sub> concentration in soil air) are not known. To ascertain this correlation was the main intention of this study.

Loess from Nussloch (SW-Germany) was chosen instead of soil because it contains solely primary CaCO<sub>3</sub>, has high carbonate content (30%) and nearly no organic carbon. Within metallic tubes, air containing increasing CO<sub>2</sub> concentrations (400, 5000 and 50000 ppm) labeled with <sup>14</sup>CO<sub>2</sub> was applied to the loess. After increasing time periods (3 days, 2 weeks, 2 months), the <sup>14</sup>C activity in loess carbonate, dissolved inorganic carbon (DIC), and CO<sub>2</sub> were measured by liquid scintillation counting.

The maximal <sup>14</sup>C activity was recovered in loess CaCO<sub>3</sub>, followed by <sup>14</sup>C activity in DIC, and the minimal <sup>14</sup>C activity was in CO<sub>2</sub>. <sup>14</sup>C activity in loess CaCO<sub>3</sub> and <sup>14</sup>C specific activity of input CO<sub>2</sub>-C were used to calculate the amount of recrystallised loess carbonate. The amount of recrystallised CaCO<sub>3</sub> increased with increasing CO<sub>2</sub> concentration. After 3 days, the amount of recrystallised carbonate was 0.0002%,

0.0003% and 0.0006% of original  $\text{CaCO}_3$  for 400, 5000 and 50000 ppm, respectively. Under 5000 and 50000 ppm  $\text{CO}_2$  concentrations, an equilibrium between  $^{14}\text{C}$  in DIC and  $\text{CaCO}_3$  was not reached even after 2 months, and the increase of the  $^{14}\text{C}$  activity in the loess carbonate was stronger at 50000 than at 5000 ppm.

$\text{CO}_2$  concentration in soil air, ranging from atmospheric values in the upper horizons up to 70000 ppm in deeper horizons, plays an important role for the soil  $\text{CaCO}_3$  re-crystallisation.