



## **Formaldehyde source apportionment and photochemical simulation in the city air of Santiago, Chile**

**Y. Elshorbany** (1,2), P. Wiesen (1), J. Kleffmann (1), R. Kurtenbach (1), M. Rubio (3), E. Lissi (3), G. Villena (3), A. R. Rickard (4), M.J. Pilling (4)

(1) Department of Physical Chemistry, University of Wuppertal, Gaußstraße 20, 42119 Wuppertal, Germany, (2) Environmental Sciences Division, National Research Center, Dokki, Giza, Egypt, (3) Facultad de Química y Biología, Universidad de Santiago de Chile, Alameda L. Bernardo O'Higgins 3363, Santiago, Chile, (4) School of Chemistry, University of Leeds, Leeds, UK (elshorbany@uni-wuppertal.de / Fax: +49(202)4392757)

During the extensive summer measurement campaign carried out from the 8<sup>th</sup>-20<sup>th</sup> March, 2005, the oxidation capacity evaluation study has revealed HCHO as a main photochemical oxidation precursor contributing by about 17 % of the total radical sources in the city air of Santiago. In this work, HCHO source apportionment and photochemical simulation analysis and its main precursors have been determined. The pair O<sub>3</sub> – NO<sub>x</sub> has been used as tracer for the photochemically formed and primary emitted HCHO, respectively. The photochemical HCHO has been simulated using a photochemical box model based on the Master Chemical Mechanism, MCM (MCMv3.1, <http://mcm.leeds.ac.uk/MCM/>) constrained with simultaneous measurements of HONO, O<sub>3</sub>, CO, NO, NO<sub>2</sub>, J(O<sup>1</sup>D), J(NO<sub>2</sub>), VOCs and meteorological parameters. The photochemically formed HCHO comprises up to 70 % of the observed HCHO in the afternoon, while the primary HCHO comprised more than 90 % during the early morning rush hour. The oxidation of alkenes contributes to photochemical HCHO formation by 70 % followed by aromatics 17% and alkanes 11 %. Oxidation of isoprene contributes alone by 23 %. The contribution of each oxidant (OH, O<sub>3</sub> and NO<sub>3</sub>) to photochemical HCHO formation has been determined. The OH is the major oxidant responsible to about 85 % of the total HCHO produced by the oxidation of the

hydrocarbons followed by O<sub>3</sub>, 14 %, while the NO<sub>3</sub> contributions are negligible.