



## **The reactions $\text{IO} + \text{NO}_3 \rightarrow \text{OIO} + \text{NO}_2$ , and $\text{I} + \text{NO}_3 \rightarrow \text{IO} + \text{NO}_2$ , - rate coefficients and product yields by LIF detection of IO**

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The technique of pulsed laser photolysis coupled to LIF detection of IO was used to study the reactions (R1a)  $\text{IO} + \text{NO}_3 \rightarrow \text{OIO} + \text{NO}_2$  and (R2)  $\text{I} + \text{NO}_3 \rightarrow$  (products) at ambient temperature. The reaction (R1) was observed for the 1<sup>st</sup> time in the laboratory, and a lower-limit for the rate coefficient of  $k_{1a} > 7 \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  is reported, which may explain anomalous ratios of OIO:IO observed in the night-time marine boundary layer. Other important results included the identification of IO (*via* laser excitation spectra around  $\lambda = 445 \text{ nm}$ ) as the product of (R2) at approximately unity yield, and  $k_2 = (1.0 \pm 0.4) \times 10^{-10} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ . The implications of these results for the night-time chemistry of the marine boundary layer are discussed.