



Spring CO₂ dynamics within sea ice: abiotical versus biological control.

B. Delille (1), A.V. Borges (1), D. Lannuzel (2), S. Becquevort (2), V. Schoemann (2), C. Lancelot (2), J.T.M. De Jong (3), B. Tilbrook (4,5), D. Delille (6), and J.-L. Tison (7)

(1) Chemical Oceanography Unit, Université de Liège, Belgium, (2) Ecologie des Systèmes Aquatiques, Université Libre de Bruxelles, Belgium, (3) DSTE, Université Libre de Bruxelles, Belgium, (4) CSIRO Marine Research, Australia, (5) ACE CRC and Antarctic Division, University of Tasmania, Australia, (6) Observatoire Océanologique de Banyuls, Université P. et M. Curie, France, (7) Glaciology Unit, Université Libre de Bruxelles, Belgium (bruno.delille@ulg.ac.be)

High latitude oceans are major sinks for atmospheric carbon dioxide (CO₂) but so far the extensive sea ice cover has been considered inert with respect to gas exchange with the atmosphere. There are growing observations that sea ice exchanges CO₂ directly with the atmosphere, highlighting the need in understanding CO₂ dynamics within sea ice. In spring ice, at least 6 processes can affect CO₂ dynamics, namely: (1) temperature change and (2) related melting of ice crystal, (3) biological activity, (4) dissolution of carbonate minerals, (5) internal convection and (6) air-ice gases exchanges. Each process can significantly impact inorganic carbon dynamics with opposing and compensating effects. To explore the relationships between sea ice-specific biogeochemical processes and spring CO₂ dynamics, we carried out three surveys in Antarctic land fast sea ice, and first year and multiyear pack ice. We tried to assess the role played by none-transport processes and give some clues of the impact of the others.

In sea ice, intense growth of microalgae is commonly observed and acts to significantly decrease the partial pressure of CO₂ (pCO₂) and promote the uptake of atmospheric CO₂ by Antarctic sea ice. Biological mediated decrease of pCO₂ is significantly enhanced by superimposed none-transport abiotic processes (temperature change and related melting of ice crystals, dissolution of carbonate minerals) while transport processes (internal convection and air-ice gas exchange) counteract the ob-

served decrease of $p\text{CO}_2$. Independent and indirect estimates of the potential CO_2 fluxes driven by these processes are consistent with measurements of air-ice CO_2 exchanges.