



The effect of NO₂ on particle formation during ozonolysis of d-limonene and α -pinene

J. K. Nøjgaard (1), C. Stenby (2), M. Bilde (3), O. J. Nielsen (3) and P. Wolkoff(1)

(1)National Institute of Occupational Health, Indoor Air Chemistry Group, Lersø Parkallé 105, DK-2100 Copenhagen Ø, Denmark.

(2) Max Planck Institute for Chemistry, Atmospheric Chemistry Division, PO Box 3060, D-55020 Mainz, Germany.

(3) University of Copenhagen, Department of Chemistry, Physical Chemistry, University of Copenhagen, Universitetsparken 5, DK-2100 Copenhagen Ø, Denmark.

(contact: e-mail: jkn@ami.dk; fax +45 39165201)

Oxidation of monoterpenes by ozone (O₃) may to be a source of submicron particles in indoor air (Long et al., 2000; Wainmann et al., 2000; Weschler and Shields, 2003). Abundant monoterpenes indoors are d-limonene and α -pinene (Wolkoff et al., 2000). Several parameters influence these processes, e.g. nitrogen oxides have previously shown an effect on particle formation from ozonolysis of β -pinene in the presence of sunlight (Pandis et al., 1991). In the present study, the particle formation during ozonolysis of α -pinene and d-limonene was studied at low relative humidity (RH<1%) in 1000 liters Tedlar bags at 294±2 K and ambient pressure. Particle number concentrations, total particle volumes and size distributions were measured using a Differential Mobility Particle Sizer - Condensation Particle Counter (DMPS-CPC).

At 50 parts per billion (ppb) monoterpene and variable O₃, the particle number and -volume increased non-linearly with the O₃ concentration and leveled off at ca. 100 ppb O₃ after 60 min of reaction. The particle number concentration was ca. 6 times larger for d-limonene than α -pinene at 50 ppb O₃, and also with a faster onset. As little as 6 ppb O₃ produced more than 10⁴ particles cm⁻³ during limonene ozonolysis, whereas 18 ppb O₃ produced only 10% during α -pinene ozonolysis.

A new series of experiments were conducted using 50 ppb of monoterpene and O₃,

and variable concentrations of nitrogen dioxide (NO₂). A small reduction in particle volume and -number was observed in the d-limonene experiments as NO₂ was added to the mixtures. The presence of NO₂ introduced an additional loss term for O₃, resulting in formation of the nitrate radical; O₃ + NO₂ → NO₃ + O₂. This affected the particle formation, since the nucleation potential of NO₃ is much lower than O₃ with respect to α-pinene and d-limonene (Bonn and Moortgat, 2002). Modeling showed that the observed decrease in particle number could be ascribed to the O₃/NO₂ reaction. In the α-pinene experiments, particle volume and -number were substantially reduced, resulting in a complete suppression of particle formation at 450 ppb NO₂. Moreover, the onset of particle formation was further delayed as the NO₂ concentration was increased. The particle size distribution appeared to be unchanged. The observed effect could not be explained by the additional loss of O₃ from the O₃/NO₂ reaction, but may be due to reaction of NO₂ with precursors of the particle formation, e.g. Criegee intermediates (Bonn et al., 2002).

The results indicate that in concentrations of monoterpenes and O₃ typical of indoor environments, d-limonene may form new particles as supported by other studies (Weschler and Shields, 2003; Wainmann et al., 2000). α-Pinene ozonolysis is not likely to produce new particles indoors (this study, Berndt et al., 2003). Furthermore, NO₂ appears to inhibit particle formation from α-pinene, even in lower ppb concentrations, whereas d-limonene is less affected.

The effect of relative humidity and identification of reaction products are subject to further studies.

LITERATURE

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