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Hygroscopic properties of sodium chloride nanoparticles

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Salt particles deliquesce at a range of relative humidity values to become solution droplets of increased size. The deliquescence relative humidity (DRH) of particles larger than the nano-size regime is only a function of their chemical properties and not of their size. Nanoparticles, on the other hand, do not have the same behavior as has been shown by a small number of experimental studies (Hämeri, Väkevää et al. 2000; Hämeri, Laaksonen et al. 2001) and theoretical calculations (Mirabel, Reiss et al. 2000; Djikaev, Bowles et al. 2001; Russell and Ming 2002). This small number of experimental studies and the contradicting theoretical predictions underline the need for further experimental evidence.

This paper presents measurements of the hygroscopic behavior of sodium chloride particles in the size range of 6-60 nm. In particular, because of much discussion in the literature concerning, on the one hand, the challenges of preparing pure nano-particles and, on the other hand, the importance of the relationship between DRH and purity, we generate NaCl nanoparticles by two independent methods.

The experimental apparatus consists of two nano-DMAs (TSI Model 3085), a nanoparticle CPC (TSI Model 3025A), and a series of Nafion-tube humidity exchangers. Polydisperse sodium chloride aerosol samples at low relative humidity ($RH < 5\%$), generated either by an electrospray generator (TSI Model 3480) or by the evaporation/condensation approach, are passed through a 210Po neutralizer and the first nano-DMA of the apparatus. The resulting monodisperse aerosol is then directed through the Nafion-tube humidity exchanger where its RH can be accurately adjusted. The size distribution of the hydrated aerosol is measured by the second nano-DMA and the nanoparticle CPC.

The results show that sodium chloride nanoparticles deliquesce at higher relative humidity values, compared to particles larger than 20-nm mobility diameter. Moreover, the hygroscopic growth factors of the nanoparticles are lower than those of the larger particles and depart from the thermodynamic predictions. These results are compared with theoretical predictions available in the literature. Importantly, the behavior of the nanoparticles generated by the two different and independent methods (viz. electrospray and evaporation/condensation) agree very well with each other, which rules out the possibility that impurities affect our measurements.

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