

Geophysical Research Abstracts, Vol. 7, 05313, 2005  
SRef-ID: 1607-7962/gra/EGU05-A-05313  
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Chemical and physical changes of organic particles upon reaction with ozone

**Y. Katrib** (1), S. T. Martin (1), H. M. Hung (1), Y. Rudich (2), P. Davidovits (3), J. T. Jayne (4), and D.

(1) Division of Engineering and Applied Sciences, Harvard University, Cambridge MA, 02138, USA,

Heterogeneous oxidation reactions are an important process in the aging of the organic fraction in atmospheric particles, specifically in affecting their hygroscopic properties and their activity as cloud condensation nuclei. Our study provides information on the heterogeneous reactions of coated polystyrene latex (PSL) coated with oleic acid layers (2-30 nm) and pure oleic acid droplets with ozone at low relative humidity. The condensed-phase products of the reacted coated particles are characterized in situ with an aerosol mass spectrometer (AMS) and a scanning mobility particle sizer (SMPS). Attenuated total reflectance infrared spectroscopy (ATR-FTIR), gas chromatography-mass spectrometry (GC-MS), and liquid chromatography-mass spectrometry (LC-MS) are further used for product identification when studying pure oleic acid droplets.

The major products detected by the AMS are 9-oxononanoic acid (20-35%) and large (>C18) oxygenated organics (35-50%) as a result of the reaction of Criegee intermediates with the oleic acid double bond. 9-oxononanoic acid (1-3%) and nonanoic acid (1-3%) are minor products. The condensed-phase products detected by GC-MS and LC-MS for pure oleic acid droplets are 1-nonanal (30%), 9-oxononanoic acid (14%), nonanoic acid (7%), octanoic acid (1%), azelaic acid (6%), and unidentified products. FTIR results show that at least some of the unidentified products contain an ester group. Additionally, the mass spectra show that at least some of the unidentified products have high molecular weight (>1000 amu), which implicates a polymerization reaction. Moreover, the observed steps of 172 amu (9-oxononanoic acid) and 188 amu (azelaic acid Criegee intermediate) in the mass spectra of high molecular weight species, provide evidence of these species as chain propagation products. The reactive uptake coefficient is determined for the coated particles. The apparent uptake coefficient decreases with increasing layer thickness and tends towards 10<sup>-3</sup>. A change in the physical behavior is observed for particles with diameters of 8-10 nm which may represent the diffusion-reaction length. Surface and bulk reactions may be distinguished. The surface reaction is found to be much faster than the bulk reaction.

Our results indicate a change in oxidation state upon O<sub>3</sub> reaction. An increase in the carbon-normalized oxygen content (z/x) in the average chemical composition of the layer is observed from 0.1 for pure oleic acid to 0.2 after high ozone exposure and is correlated with an increase in particle density. After reaction, the droplets take up water at lower relative humidities compared to the unreacted droplets as shown by the environmental scanning electron microscopy (ESEM) study. The increased hygroscopic response may indicate that the aging of atmospheric organic aerosol particles has significant impact on radiative forcing.