Geophysical Research Abstracts, Vol. 10, EGU2008-A-07892, 2008 SRef-ID: 1607-7962/gra/EGU2008-A-07892 EGU General Assembly 2008 © Author(s) 2008



Facing the $^{15}{ m N}$ tracer technique with the process of codenitrification

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It was first reported by Tanimoto et al. 1 and Shoun et al. 2 that beside the commonly known denitrification process N2O and N2 can be also produced by a unique reaction of soil fungi species, a process which they called codenitrification. During this reaction both gaseous N compounds evolve thereby as hybrid species, where two N atoms of two different educts will be combined to form either N2O or N2 (i.e. NH4 and NO_2^-)^{1,2}. Laughlin & Stevens³ recently showed that at least for microbial soil N_2 release (acid brown earth; Typic Dystrochrept) codenitrification might account for up to 92 %. However, a simultaneous production of N₂O or N₂ in soils due to denitrification and codenitrification violates the assumptions of current ¹⁵N based mathematical approaches⁴⁻⁷. Thus, novel ¹⁵N based equations were derived to permit a calculation of soil N₂ and N₂O release in the course of both processes. The calculation is based on a single isotopic analysis of a sampled N₂ or N₂O mixture and the determination of the ¹⁵N abundance of nitrite and nitrate (simplified approach) or of ammonium, nitrite, and nitrate (comprehensive approach). Calculations are even processable under conditions where all basal nitrogen educts (ammonium, nitrite, and nitrate) are enriched in ¹⁵N. An additional determination of concentrations of dissolved N compounds is unnecessary.

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